

ON 7:8-DIMETHOXYISOCARBOSTYRIL. A CASE OF TRANS-  
ESTERIFICATION INVOLVING AN AROMATIC ALKOXYL GROUP

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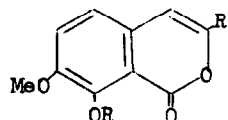
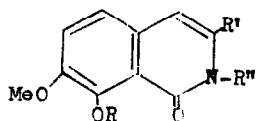
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D. Bain, W. H. Perkin (Jun.) and R. Robinson<sup>1</sup> obtained a compound, m.p. 233° by heating 7:8-dimethoxyisocarbostyril-3-carboxylic acid (I) and regarded it as 7:8-dimethoxyisocarbostyril (II). The compound was not analysed and only recently L. I. Linevich<sup>2</sup> also regarded it as (II). The compound which is insoluble in sodium hydrogen carbonate gives a blue ferric reaction and analysed for  $C_{10}H_5O_2N(OCH_3)_2$ . In chloroform solution it showed i.r. absorption at  $3330\text{ cm}^{-1}$  and  $1730\text{ cm}^{-1}$  (associated OH and ester CO). On methylation with methyl iodide and potassium carbonate in acetone solution, the compound afforded N-methyl-7:8-dimethoxyisoquinolone-3-carboxylate (III), m.p. 160-61° (negative ferric reaction) which was also

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1. D. Bain, W. H. Perkin (Jun.) and R. Robinson, J. Chem. Soc. 105, 239 (1914).
  2. L. I. Linevich, Zhur. obshchei khim. 26, 2510-14 (1958).

obtained by the methylation of methyl 7:8-dimethoxyisocarbostyryl-3-carboxylate (IV). The compound m.p. 233°



- I, R=Me, R' = COOH, R'' = H  
 II, R=Me, R' = R'' = H  
 III, R=Me, R' = COOMe, R'' = Me  
 IV, R=Me, R' = COOMe, R'' = H  
 V, R=H, R' = COOMe, R'' = H  
 VI, R=H, R' = COOH, R'' = H  
 VII, R=R' = R'' = H

- VIII, R=R' = H  
 IX, R=Me, R' = H  
 X, R=Me, R' = COOH  
 XI, R=Me, R' = COOMe  
 XII, R=H, R' = COOMe

must be regarded as methyl 7-methoxy-8-hydroxyisocarbostyryl-3-carboxylate (V) which was further substantiated by hydrolysis with aqueous potassium carbonate to the related acid (VI), m.p. 315° followed by decarboxylation with copper bronze-quinoline to 7-methoxy-8-hydroxyisocarbostyryl (VII), m.p. 209-10°. The latter was synthesised independently from 7-methoxy-8-hydroxyisocoumarin (VIII), m.p. 450-51° and ammonia.

It is pertinent to point out that P. K. Banerjee and D. N. Choudhury<sup>3</sup> have recently described a synthesis of 7:8-dimethoxyisocoumarin (IX), m.p. 149-50° from 7:8-dimethoxy-3:4-dihydroisocoumarin by bromination with N-bromosuccinimide followed by dehydrobromination.

3. P. K. Banerjee and D. N. Choudhury, J. Indian Chem. Soc. 41, 221 (1964).

C. Ribbens, c. Van der Stelt and W. Th. Nauta<sup>4</sup> reported melting point 105.5-107° for the same compound obtained by the decarboxylation of 7:8-dimethoxyisocoumarin-3-carboxylic acid (X). A careful reinvestigation of this decarboxylation has now led to the isolation of four products, viz., 7:8-dimethoxyisocoumarin (IX), m.p. 115°, 7-methoxy-8-hydroxyisocoumarin (VIII), m.p. 150-51°, methyl 7:8-dimethoxyisocoumarin-3-carboxylate (XI), m.p. 174-76° and methyl 7-methoxy-8-hydroxyisocoumarin-3-carboxylate (XII), m.p. 194°. P. K. Banerjee and D. N. Choudhury's product was obviously (VIII) arising from demethylation of 8-methoxyl group. The authentic 7:8-dimethoxyisocoumarin was converted into 7:8-dimethoxyisocarbostyryl (II), m.p. 178-79° and characterised by the picrate, m.p. 150-51°.

The formation of above esters in the isocarbostyryl and isocoumarin series is unique and it is obviously a case of intermolecular trans-esterification involving an aromatic methoxyl group. The methoxyl group at the 8-position is labile on account of the participation of the neighbouring carbonyl group.

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4. C. Ribbens, c. Van der Stelt and W. Th. Nauta,  
Rec. Trav. chim. **79**, 78 (1960).